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1 (Sem-4/FYUGP) PHY 42 MJ

2026

PHYSICS

(Major)

Paper : PHY4400204MJ

(Quantum Mechanics-I)

Full Marks : 45

Time : 2 hours

*The figures in the margin indicate
full marks for the questions.*

1. Give very short answers the following :

$$1 \times 5 = 5$$

- (a) Write down the dimension of Planck's constant.
- (b) Name an experiment that suggests wave nature of radiation.
- (c) Write down Schrödinger equation for a free particle.
- (d) What is Hilbert space?
- (e) What is a quantum dot?

2. Answer the following questions : **(any five)**

$$2 \times 5 = 10$$

- (a) Write down *any one* limitation of classical mechanics in explaining photoelectric effect.
- (b) Find the ratio of the de-Broglie wavelength to the Compton wavelength of a particle.

(c) Is it possible to simultaneously specify the kinetic energy and linear momentum of a mechanical system? Justify your answer.

(d) A nucleon is confined to a nucleus of radius $5 \times 10^{-15} m$. Calculate the minimum uncertainty in the momentum of the nucleon.

(e) State the physical significance of the wave function $\psi(x, t)$?

(f) What is a wave packet? How it affect the localization of the particle?

(g) Distinguish between phase velocity and group velocity.

(h) Justify that tunnelling is a quantum mechanical phenomenon.

(i) Show that eigenvalues corresponding to Hermitian operators are real.

- (j) Distinguish between a classical and a quantum harmonic oscillator.

3. Answer **any four** questions : $5 \times 4 = 20$

- (a) Calculate the minimum energy required by a photon to transfer half of its energy to an electron at rest.
- (b) Find the condition for two wave functions to be orthonormal. Find the probability current density for the wave function

$$\psi = \frac{e^{ikr}}{r} \quad 2\frac{1}{2} + 2\frac{1}{2} = 5$$

- (c) What do you mean by expectation value of a dynamical variable? Find the expectation value of momentum for a particle described by the wavefunction

$$\psi(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{\pi x}{L}\right) \text{ for } 0 < |x| < L$$

$$= 0 \text{ elsewhere} \quad 2 + 3 = 5$$

- (d) Show that if $\psi_1(\vec{r})$ and $\psi_2(\vec{r})$ are two independent solutions of the Schrödinger equation then

$\psi(\vec{r}) = a_1\psi_1(\vec{r}) + a_2\psi_2(\vec{r})$ is also a solution of Schrödinger equation.

- (e) Write the significance of energy time uncertainty principle. Discuss how we can estimate the range of interaction for fundamental force using energy time uncertainty relation and virtual particles.

- (f) Explain *two* slit experiments with electron and discuss its significance.

- (g) Discuss the properties of a Gaussian wave packet.

- (h) Write down the radial wave function for 1s state of hydrogen atom. Also, compare the probabilities of a 1s electron in the hydrogen atom being at a distance a_0 from the nucleus than at a distance $a_0/2$. $1 + 4 = 5$

4. Answer the following questions : (any one)
10×1=10

(a) What was the main discrepancy of Rayleigh-Jeans formula for black body radiation? Derive Planck's law and discuss how the discrepancy was solved.
2+6+2=10

(b) A particle is in an infinite potential well

$$V(x) = \begin{cases} 0, & 0 < x < a \\ \infty, & \text{elsewhere} \end{cases}$$

Find out the energy eigenvalues and eigenfunctions. Also schematically show the wave functions and probability densities for first three states.

8+2=10

(c) Show that average motion of a wave packet follows Newton's second law of motion. What is the significance of Ehrenfest's theorem.

(d) (i) Explain the significance of the quantum numbers n , l and m . Can the principal quantum number of electron in hydrogen atom be zero.

3+1=4

(ii) Show that the most probable distance of the electron from the nucleus in the ground state of hydrogen atom is equal to the Bohr's radius a_0 . 6